TRICHOTHECENE METABOLISM STUDIES. 2. STRUCTURE OF 3a-(1"B-D-GLUCOPYRANOSIDURONYL)-&-ISOVALERYLOXY-SCIRPEN-3,4B, 15-TRIOL 15-ACETATE PRODUCED FROM T-2 TOXIN IN VITRO

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## Abstract

The preparation and structure determination of the title compound is described.

The epoxytrichothecene mycotoxins are a large family of fungal metabolites that exhibit a range of impressive biological activities.  $^3$  They are potent inhibitors of protein synthesis in eukaryotes and many have been implicated in diseases of plants, animals, and humans.  $^{3a}$  In spite of the toxicological significance of these compounds, however, relatively little is known about their metabolic fate in mammalian systems.  $^{4,5}$  We recently described the  $\frac{in}{i}$  vitro production of glucuronide 2 from anguidine (1) and provided the first concrete evidence that

glucuronidation may be a significant pathway for the trichothecene metabolism  $\frac{in}{in} \frac{vivo}{vivo}$ . Questions remained, however, about the generality of this conclusion. Consequently we have initiated a parallel series of experiments using T-2 toxin (3) and report herein the enzymatic production and structure determination of glucuronide  $\frac{4}{i}$  as the major in vitro T-2 conjugate. 6

[ $^3$ H] T-2 toxin (4.7 nM, 0.026 μCi)  $^7$  was incubated with uridine 5'-diphosphoglucuronic acid (UDPGA, 12 mM), β-naphthoflavone-induced hepatic microsomes from male Long-Evans rats (1.2 mg protein/mL),  $^8$  MgCl $_2$  (10 mm), and phosphate buffer (10 mM, pH 7.7) at 37°C. HPLC analysis of the mixture after 2 h indicated the presence of a new product (49%) glucuronide  $^4$  (R $_t$  15.5 min).  $^9$ ,  $^{10}$  The same product was produced by incubation of T-2 toxin (150 μM) with [ $^1$ 4C] UDPGA (0.7 μM, >180 mCi/mmole) using the protocol outlined above. Scale-up of this procedure  $^{11}$  using unlabelled T-2 toxin followed by HPLC purification  $^{10}$  afforded pure  $^4$ 12 (10% isolated yield).

The FAB mass spectrum  $^{13}$  of  $^{4}$  is consistent with a 1:1 adduct of HT-2 toxin (5) and glucuronic acid [m/e 645 (MNa $_{2}^{+}$ ), 623 (MNa $_{2}^{+}$ ), 601 (MH $_{2}^{+}$ ), 389 (M $_{2}^{+}$ -C<sub>6</sub>H<sub>9</sub>O<sub>7</sub>-H<sub>2</sub>O), 307 (M $_{2}^{+}$ -C<sub>6</sub>H<sub>9</sub>O<sub>7</sub>-C<sub>5</sub>H<sub>9</sub>O<sub>2</sub>)], a conclusion supported by  $^{1}$ H NMR data which showed a single acetyl resonance.  $^{12}$  Comparison of the  $^{1}$ H signals for H-4 ( $^{6}$  4.58), H-3 (4.44) and H-15 (4.30 and 3.99) of  $^{4}$  measured in MeOH-d<sub>4</sub> with those of authentic HT-2 toxin ( $^{5}$ ), ( $^{6}$  4.38, H-4; 4.10, H-3; 4.27 and 4.00, H-15) suggested that  $^{4}$  is a glucuronide derivative of  $^{5}$ .  $^{14}$  The linkage between the trichothecene nucleus and glucuronic acid was determined by conversion to the peracetate methyl ester derivative  $^{6}$ . This compound was identical to an authentic sample synthesized from T-2 toxin ( $^{3}$ ) and bromosugar  $^{7}$  by a Koenigs-Knorr reaction (20% yield; 94% based on recovered T-2 toxin). Thus, the structure of this metabolite is correctly described by formula  $^{4}$ .

This experiment shows that T-2 toxin or its well-known hydrolysis product HT-2 (5) are viable substrates for microsomal glucuronyl transferase. Based on our previous experience with anguidine, it is likely that T-2 toxin is hydrolyzed to HT-2 before conjugation with UDP-glucuronic acid. A Nevertheless, it is now reasonable to speculate that glucuronic acid conjugates of T-2 metabolites will be produced in vivo. Structural studies of in vivo metabolites will be reported in due course. 17

## References

- 1. (a) Department of Chemistry; (b) Department of Applied Biological Sciences.
- 2. Holder of the Roger and Georges Firmenich Career Development Chair in Natural Products Chemistry, 1981-84; Fellow of the Alfred P. Sloan Foundation, 1982-86.
- (a) "Developments in Food Science, Vol. 4; Trichothecenes: Chemical, Biological, and Toxicological Aspects"; Ueno, Y., Ed.; Elsevier: New York, 1983. (b) Doyle, T.W.; Bradner, W.T. In "Anticancer Agents Based on Natural Product Models"; Cassady, J.M.; Douros, J.D., Eds.; Academic Press: New York, 1980; Chapter 2. (c) Ueno, Y. Adv. Nutr. Res. 1980, 3, 301. (d) Tamm, C. Fortschr. Chem. Org. Naturst. 1974, 31, 63. (e) Bamburg, J.R.; Strong, F.M. In "Microbial Toxins"; Kadis, S., Ciegler, A., Ajl, S.J., Eds.; Academic Press: New York, 1971; Vol. 7, p 207.
- Roush, W.R.; Marletta, M.A.; Russo-Rodriguez, S.; Recchia, J. <u>J. Am. Chem. Soc.</u> 1985, 107, 3354.
- (a) Yoshizawa, T.; Sakamoto, T.; Okamoto, K. <u>Appl. Environ. Microbiol. 1984, 47</u>, 130.
  (b) Yoshizawa, T.; Sakamoto, T.; Anyano, Y.; Mirocha, C.J. <u>Agric. Biol. Chem. 1982</u>, 46, 2613. (c) Robison, T.S.; Mirocha, C.J.; Kurtz, H.J.; Behrens, J.C.; Weaver, G.A.; Chi, M.S. <u>J. Agric. Food Chem. 1979</u>, 27, 1411. (d) Yoshizawa, T.; Takeda, H.; Ohi, T. Agric. Biol. Chem. 1983, 47, 2133. (e) King, R.R.; McQueen, R.E.; Levesque, D.; Greenhalg, R. <u>J. Agric. Food Chem. 1984</u>, 32, 1181. (f) Yoshizawa, T.; Swanson, S.P.; Mirocha, C.J. <u>Appl. Environ. Microbiol. 1980</u>, 39, 1172. (g) Yoshizawa, T.; Swanson, S.P.; Mirocha, C.J. <u>Ibid. 1980</u>, 40, 901. (h) Matsumoto, H.; Ito, T.; Ueno, Y. <u>Jpn. J. Exp. Med. 1978</u>, 48, 393. (i) Ohta, M.; Matsumoto, H.; Ishii, K.; Ueno, Y. <u>J. Biochem. (Tokyo)</u> 1978, 84, 697.
- 6. The only metabolic transformations of  $\underline{3}$  documented prior to our work had been deacylation reactions catalyzed by microsomal esterases,  $5^{f-1}$  and the P<sub>450</sub> mediated hydroxylation of the 3'-position. 5a, b
- We thank Dr. K. Hunter (Uniformed Services University of Health Sciences, Dept. of Defense, Bethesda, Maryland) for a generous sample of [3H] T-2 toxin.
- 8. Ryan, D.; Lee, A.Y.H.; Levin, W. In "Methods in Enzymology"; Fleischer, S., Packer, L., Eds.; Academic Press: New York, Vol. 52, pp 117-123.
- The microsomes were removed by centrifugation at the end of the incubation. The products were concentrated on a C18 Sep-Pak cartridge (Waters Assoc.) and then were separated by HPLC.
- 10. The  $\mu$ -Bondapak C18 column (3.9 mm x 30 cm, Waters Assoc.) was used for all analyses and isolations (100% H<sub>2</sub>O for 2 min, 0-45% MeOH linear ramp for 15 min., 45-60% MeOH linear ramp for 15 min., 1.5 mL/min).
- 11. UDPGA (12 mM), T-2 toxin (738 µM) and 1.22 mg/mL of microsomal protein (Sprague-Dawley rats) were incubated for 3.5 h.
- 12. Data for 4:  $^1$ H NMR (MeOH-d<sub>4</sub>, 250 MHz)  $^{\delta}$  5.75 (d, 1 H, J=5.6 Hz, H-8), 5.31 (d, 1 H, J=5.0 Hz, H-10), 4.68 (d, 1 H, J=7.6 Hz, H-1"), 4.58 (d, 1 H, J=3.0 Hz, H-4), 4.44 (dd, 1 H, J=3.1, 4.8 Hz, H-3), 4.30 (d, 1 H, J=12.4 Hz), 4.23 (d, 1 H, J=5.8 Hz, H-11), 3.99 (d, 1 H, J=12.4 Hz, H-15B), 3.68 (d, 1 H, J=4.9 Hz, H-2), 3.5-3.2 (m, partially obscured by solvent peak, sugar), 2.96 (d, 1 H, J=4.1 Hz, H-13A), 2.77 (d, 1 H, J=4.0 Hz), 2.38 (dd, 1 H, J=5.6, 15 Hz, H-7A), 2.05 (s, 3 H, -0Ac), 1.73 (s, 3 H, H-16), 1.00 (m, 6 H, H-4' isovalery1), 0.82 (s, 3 H, H-14); FAB mass spectrum (glycerol dispersion) m/e 645

- $(MNa_2^+)$ , 623  $(MNa^+)$ , 601  $(MH^+)$ , 389  $(M^+-c_6H_90_7-H_20)$ , 307  $(M^+-c_6H_90_7c_5H_90_2)$ .
- 13. The FAB mass spectral measurements were performed by Dr. C. Costello and S. Maleknia using the facility supported by NIH Research Grant #RR00317 from the Biotechnology Resources Branch, Division of Research Resources (Principal Investigator: Prof. K. Biemann).
- 14. The corresponding chemical shifts for H-4 (δ 5.70), H-3 (δ 4.23) and H-15 (4.32 and 4.10) of T-2 toxin (3) indicate that H-4 is not acylated in metabolite 4.
- 15. Data for 6: mp 82-84°C;  $[\alpha]_D^{20}$  3.3° (c=0.61, CHCl<sub>3</sub>); H NMR (CDCl<sub>3</sub>, 250 MHz)  $\delta$  5.88 (d, 1 H, J=2.7, H-4), 5.74 (d, 1 H, J=5.8 Hz, H-8), 5.25 (m, 3 H, H-8 and 2 sugar H's), 5.10 (m, 1 H, sugar), 4.79 (d, 1 H, J=7.5 Hz, H-1"), 4.32 (dd, 1 H, J=2.6, 5.0 Hz, H-3), 4.26 (d, 1 H, J=12.7 Hz, H-15A), 4.16 (d, 1 H, J=5.9 Hz, H-11), 4.07 (d, 1 H, J=12.6 Hz, H-15B), 3.97 (d, 1 H, J=9.7 Hz, H-5"), 3.73 (d, 1 H, J=5.0 Hz, H-2), 3.71 (s, 3 H, -CO<sub>2</sub>Me), 3.04 (d, 1 H, J=3.8 Hz, H-13A), 2.77 (d, 1 H, J=3.9 Hz, H-13B), 2.36 (dd, 1 H, J=5.8, 15.2 Hz), 2.11-2.0 (5s, 15 H, -OAc), 1.74 (s, 3 H, H-16), 0.95 (m, 6 H, H-4'), 0.71 (s, 3 H, H-14); IR (CHCl<sub>3</sub>) 2950, 1750 (br), 1430, 1365, 1210 (br), 1030 (br) cm<sup>-1</sup>; FAB mass spectrum (DMSO/glycerol dispersion) m/e 783 (MH<sup>+</sup>), 317 (C<sub>13</sub>H<sub>17</sub>O<sub>9</sub><sup>+</sup>, sugar); EI mass spectrum m/e 767 (M<sup>+</sup>-CH<sub>3</sub>), 698 (M<sup>+</sup>-C<sub>5</sub>H<sub>8</sub>O), 680 (M<sup>+</sup>-C<sub>5</sub>H<sub>10</sub>O<sub>2</sub>); high resolution mass spectrum, calcd for C<sub>32</sub>H<sub>4</sub>O<sub>16</sub> (M<sup>+</sup>-C<sub>5</sub>H<sub>10</sub>O<sub>2</sub>) 680.2316; Found, 680.232 ± 0.001.
- Bollenback, G.N.; Long, J.W.; Benjamin, D.G.; Lindquist, J.A. <u>J. Am. Chem. Soc.</u> 1955, 77, 3310.
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